Tetrahedron Letters No.23, pp. 2727-2730, 1968. Pergamon Press. Printed in Great Britain.

APPLICATIONS OF INTERNAL NUCLEAR OVERHAUSER EFFECTS (NOE)

IN ORGANIC CHEMISTRY :

1,2,3,4-TETRAMETHYLPHENANTHRENE 1)

R.H.MARTIN and J.C.NOULS

Service de Chimie Organique, Fac. Sc., Université Libre de Bruxelles, 50, Av.F.D.Roosevelt - Bruxelles 5. Belgique.

(Received in UK 1 March 1968; accepted for publication 6 March 1968)

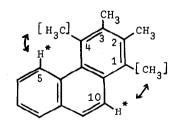
Internal nuclear Overhauser effects have been used to assign the methyl signals in the NMR spectrum of 1,2,3,4-tetramethylphenanthrene.

In 1965, Anet and Bourn²⁾ pointed out that "studies of NOE's (Nuclear Overhauser Effects) can be of considerable stereochemical and conformational interest in organic chemistry". However, relatively few applications of this technique have so far been published³⁾.

To our knowledge, no case involving an aromatic proton has yet been disclosed. We now wish to report such a case.

In 1,2,3,4-tetramethylphenanthrene ⁴⁾, saturation of the *overcrowded* methyl group $[CH_3(4)]$ should give a NOE for H(5), whereas saturation of the other CH₃'s should have no effect on this particular proton. On the other hand, saturation of CH₃(1) could have a small effect on the *peri* proton H(10).

Table 1.



 $\frac{60 \text{ Mc. spectrum (CDCl}_3)}{(CH_3(2) + CH_3(3)]} \text{ quasi singlet (6 protons) at 146 c/s (partially resolved at 100 Mc; <math>\Delta \delta_{obs.} = 2 \text{ c/s}$, $CH_3(1)$ singlet at 158 c/s, $CH_3(4)$ singlet at 173 c/s, H(10) doublet centered at 476 c/s, H(5) multiplet centered at 511 c/s.

For NOE experiments, 12.5% CDCl₃ solutions of freshly sublimed 1,2,3,4-tetramethylphenanthrene were thoroughly degazed under high vacuum and sealed under helium. These solutions were studied, at 60 Mc, by a frequency sweept method, using TMS as an internal field frequency lock. The results are summarized in Table 1 and 2.

Saturation of $CH_3(4)$ results in an enhancement of $33 \pm 3\%$ of the integrated area of the multiplet of H(5), leaving the other parts of the spectrum unaltered, whereas saturation of $[CH_3(2) + CH_3(3)]$ has no effect on the signals of the aromatic protons.

Irradiation at the $CH_3(1)$ frequency gives a small (11 \pm 3%) but apparently significant increase of the integrated area of the H(10) signals.

The assignments of the CH_3 's signals made on the basis of *internal* nuclear Overhauser effects are fully substantiated by the study of the chemical shifts, with reference to the monomethylphenanthrenes ^{5,6)}.

Irradiation	Area of the signals in arbitrary units			Enhancement in %
Indiation	* H(5)	* H(10)	H(6) + H(7) +H(8) + H(9)	(<u>+</u> 3)
None	21.5	30.0	100	-
at CH ₃ (1)	** 21.5	33.3	100	11 H(10)
at CH ₃ (2)+CH ₃ (3)	21.9	30.9	100	0
at CH ₃ (4)	28.7	30.3	100	33.5 H(5)

Table 2.

* Average of 3 runs with H(6) + H(7) + H(8) + H(9) = 100.

** Value corrected for a small beating effect.

It is clear, from these results, that NOE experiments can yield valuable informations in the field of Overcrowded Aromatic Hydrocarbons.

Further work is in progress.

Acknowledgments:

Thanks are due to the staff of JEOL CO (Europe) for technical assistance with the double irradiation experiments and to the Fonds de la Recherche Scientifique Fondamentale Collective for financial support.

References:

- Part XV of "Applications of NMR spectroscopy in the field of polycondensed aromatic systems". For Part XIV see : R.H.Martin, N.Defay, F.Geerts-Evrard and D.Bogaert-Verhoogen, <u>Tetrahedron</u>, suppl. Part I, 181 (1967).
- 2) F.A.L.Anet and A.J.R.Bourn, <u>J.Am.Chem.Soc.</u>, <u>87</u>, 5250 (1965).
- M.C.Woods, L.Miura, Y.Nakadaira, A.Terahara, M.Maruyama and K.Nakanishi, <u>Tetrahedron Letters</u>, <u>4</u>, 321 (1967).
 - M.Ohtsuru, M.Teraoka, K.Tori and Ken'ichi Takeda, <u>J.Chem.Soc.</u>(B), 1033 (1967).
 - J.G.Colson, P.T.Lansbury and F.D.Saeva, <u>J.Am.Chem.Soc.</u>, <u>89</u>, 4987 (1967).
 - J.C.Nouls, G.Van Binst and R.H.Martin, <u>Tetrahedron Letters</u>, <u>41</u>, 4065 (1967).
 - M.C.Woods, H.C.Chiang, Y.Nakadaira and K.Nakanishi, <u>J.Am.Chem.Soc.</u>, <u>90</u>, 522 (1968).
 - J.C.Nouls, P.Wollast, J.C.Braekman, G.Van Binst, J.Pécher and R.H. Martin, This Journal, accompanying communication.
- 4) C.L.Hewett and R.H.Martin, J.Chem.Soc., 1936 (1940).
 - W.Herwig, W.Metlesics and H.Zeiss, J.Am.Chem.Soc., 81, 6203 (1959).

•

- 5) P.Durand, J.Parello, N.P.Buu-Hol and L.Alais, <u>Bull.Soc.Chim.Fr.</u>, 2438 (1963).
- R.H.Martin, N.Defay, F.Geerts-Evrard and H.Figeys, <u>Bull.Soc.Chim.Belg.</u>, <u>73</u>, 199 (1964).